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The chemistry answer is “the elements.” The elements listed in the periodic table constitute all of the ordinary matter, visible and invisible, that we know about. That means that both the stars that we can’t even see with our eyes and our eyes themselves are composed of the elements.

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This should be the last course a student takes before high school biology. Typically, we recommend that the student take this course during the same year that he or she is taking prealgebra. Exploring Creation With Physical Science provides a detailed introduction to the physical environment and some of the basic laws that make it work. The fairly broad scope of the book provides the student with a good understanding of the earth's atmosphere, hydrosphere, and lithosphere. It also covers details on weather, motion, Newton's Laws, gravity, the solar system, atomic structure, radiation, nuclear reactions, stars, and galaxies. The second edition of our physical science course has several features that enhance the value of the course: \* There is more color in this edition as compared to the previous edition, and many of the drawings that are in the first edition have been replaced by higher-quality drawings. \* There are more experiments in this edition than there were in the previous one. In addition, some of the experiments that were in the previous edition have been changed to make them even more interesting and easy to perform. \* Advanced students who have the time and the ability for additional learning are directed to online resources that give them access to advanced subject matter. \* To aid the student in reviewing the course as a whole, there is an appendix that contains questions which cover the entire course. The solutions and tests manual has the answers to those questions. Because of the differences between the first and second editions, students in a group setting cannot use both. They must all have the same edition. A further description of the changes made to our second edition courses can be found in the sidebar on page 32.

In this book you will learn about the history of science, how to do science, the history of life, how your body works, and some of the amazing living creatures that exist in God's Creation.

This book begins with a lesson on the nature of astronomy, and then it covers the major structures of our solar system. Starting with the sun and working towards Pluto, the student will learn details about all nine planets (or is it eight? - your student will have to decide) in the solar system. Along the way, the student will also learn about Earth's moon, the asteroid belt, and the Kuiper belt. After that, the student will move outside our solar system and learn about the stars and galaxies that make up God's incredible universe. Finally, the student will learn about space travel and what it takes to be an astronaut! The activities and projects use easy-to-find household items and truly make the lessons come alive! They include making a solar eclipse, simulating the use of radar to determine a hidden landscape, and making a telescope. We recommend that you spend the entire school year covering this book, devoting approximately two sessions per week to the course.

Friendly Chemistry is a truly unique approach to teaching introductory chemistry. Used by home schoolers and charter, public and private school students world-wide for over ten years, Friendly Chemistry presents what is often considered an intimidating subject as a genuinely fun, enjoyable experience. Whether you're a high-school aged student needing a lab science course or a "non-traditional" student looking for a refresher course to help you prepare for an upcoming entrance exam, Friendly Chemistry can help you accomplish your goal in a "painless" way! If you do have aspirations of a future in a science field, Friendly Chemistry can give you the solid foundation you need to succeed in subsequent courses.Friendly Chemistry was written using simple language and a host of analogies to make learning (and teaching!) chemistry easy. The chemistry concepts presented in Friendly Chemistry are NOT watered-down. The concepts are just explained in ways that are readily understood by most learners. Coupled with these explanations is a host of teaching aids, labs and games which makes the learning concrete and multi-sensory. Students find the course fun and painless. Parents often comment, "I wish I had had this when I was taking chemistry. Now it all makes so much sense!" Friendly Chemistry covers the same topics taught in traditional high school chemistry courses. The course begins with an introduction to atomic theory followed by discussion of why the elements are arranged the way they are in the periodic table. Quantum mechanics comes next using the acclaimed "Doo-wop" Board as a teaching aid. Next comes a discussion of how atoms become charged (ionization), followed by an explanation of how charged atoms make compounds. The mole is introduced next, followed by a discussion of chemical reactions. Stoichiometry (predicting amounts of product produced from a reaction) is treated next followed by a discussion of solutions (molarity). The course is wrapped up with a discussion of the ideal gas laws. Please note that this is the STUDENT EDITION. Volumes 1 and 2 of the TEACHER'S EDITION must be purchased separately in order to have all materials necessary to complete this chemistry course. More information regarding Friendly Chemistry including answers to many frequently asked questions may be found at [www.friendlychemistry.com](http://www.friendlychemistry.com).

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